

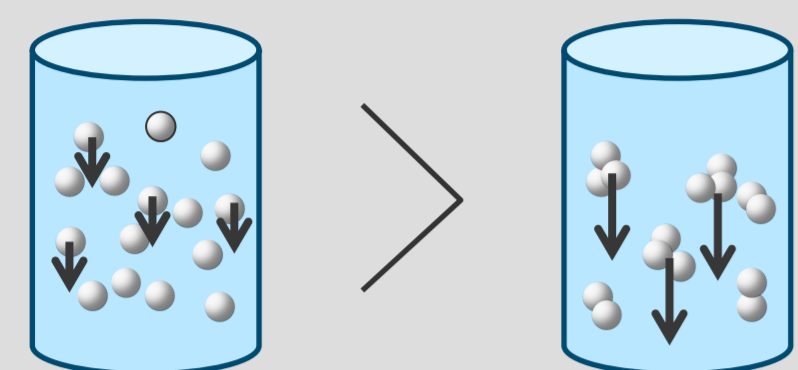
# The effect of applied electric field conditions on the aggregation of particles in aqueous slurries

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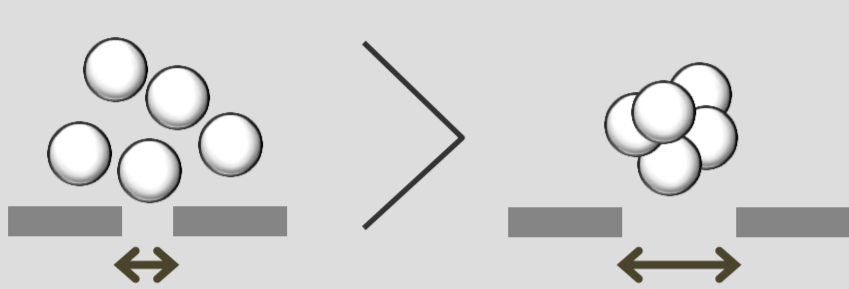
## Introduction

By forming aggregates, the efficiency of solid – liquid separation can be improved.

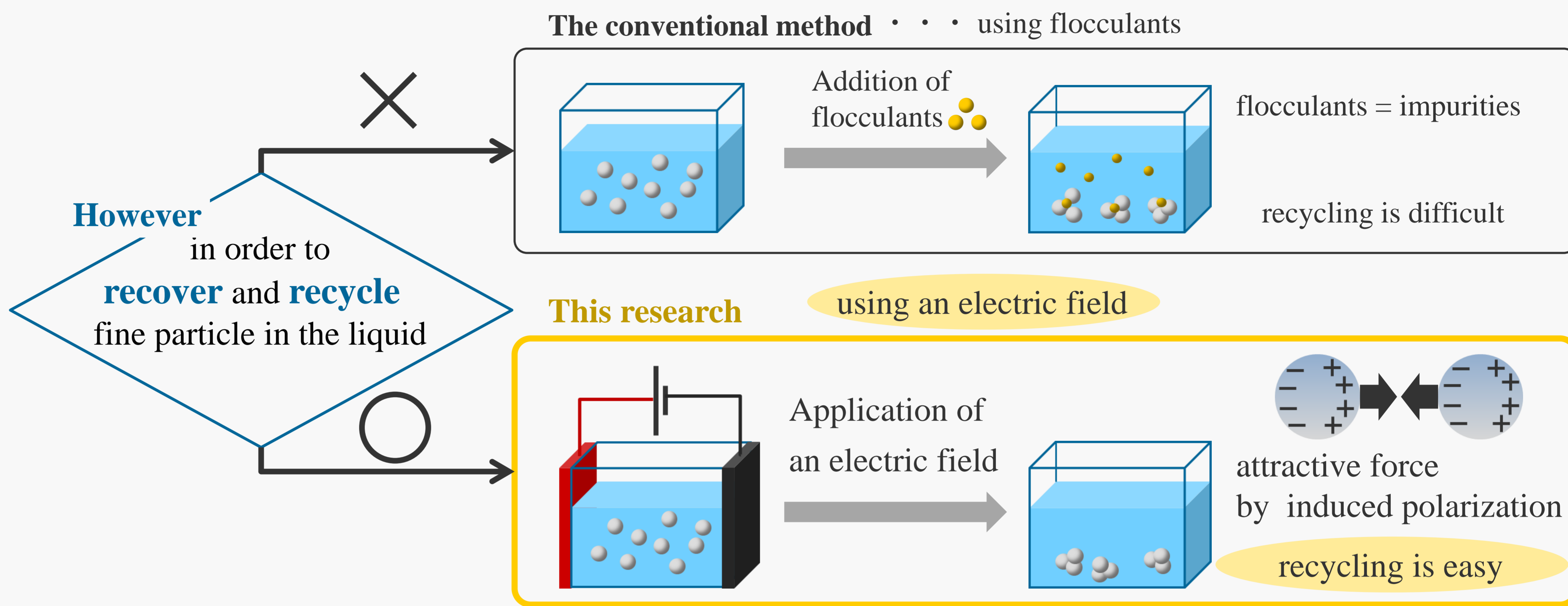
ex.1 sedimentation thickening



ex.2 filtration



- quick separation
- cost reduction



In this paper we discussed how the distance between two electrodes and the applied voltage affect on particle aggregation.

## Experiments

- powder** alumina (AES-12), average particle size = 0.48  $\mu\text{m}$
- dispersion medium** deionized water
- pH adjustment agent** nitric acid, aqueous ammonia solution

Table 1 the slurry conditions

particle concentration [vol%]	pH [-]
0.1	3.4 ~ 3.5
	7.0 ~ 7.2
	8.8 ~ 8.9
	10.4 ~ 10.5

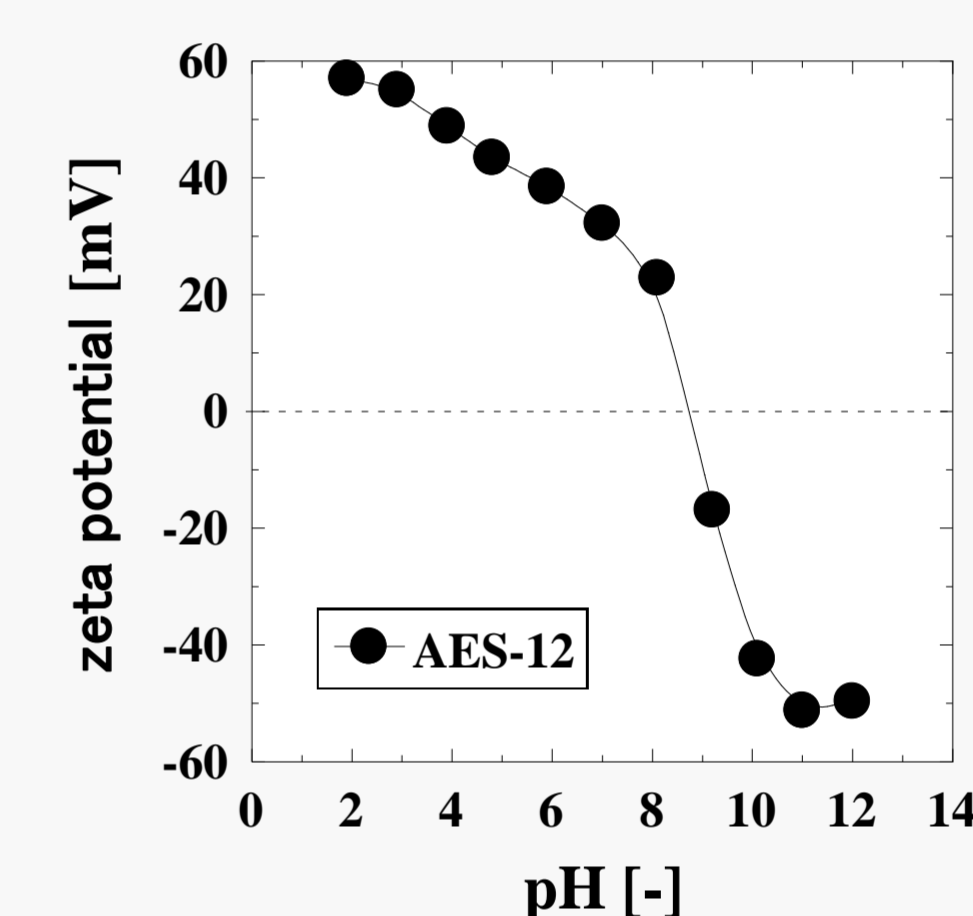


Fig.2 zeta potential

Table 2 the applied electric field conditions

electric field strength [ $\text{V} \cdot \text{m}^{-1}$ ]	125			
distance between two electrodes [mm]	40	60	80	100
voltage [V]	5.0	7.5	10.0	12.5

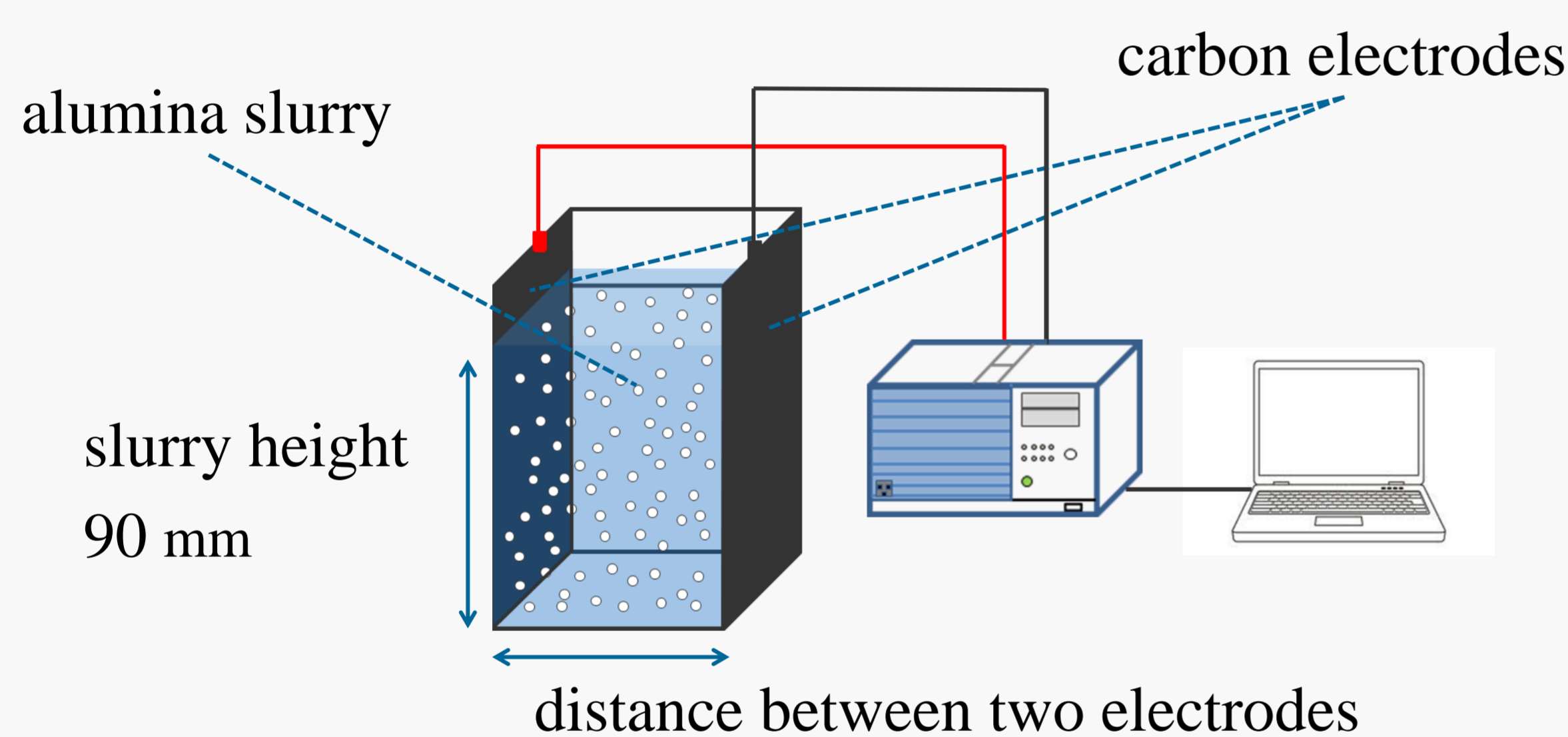


Fig.1 the experimental device

## Results and Discussion

■ observation of interface between supernatant and slurry layer

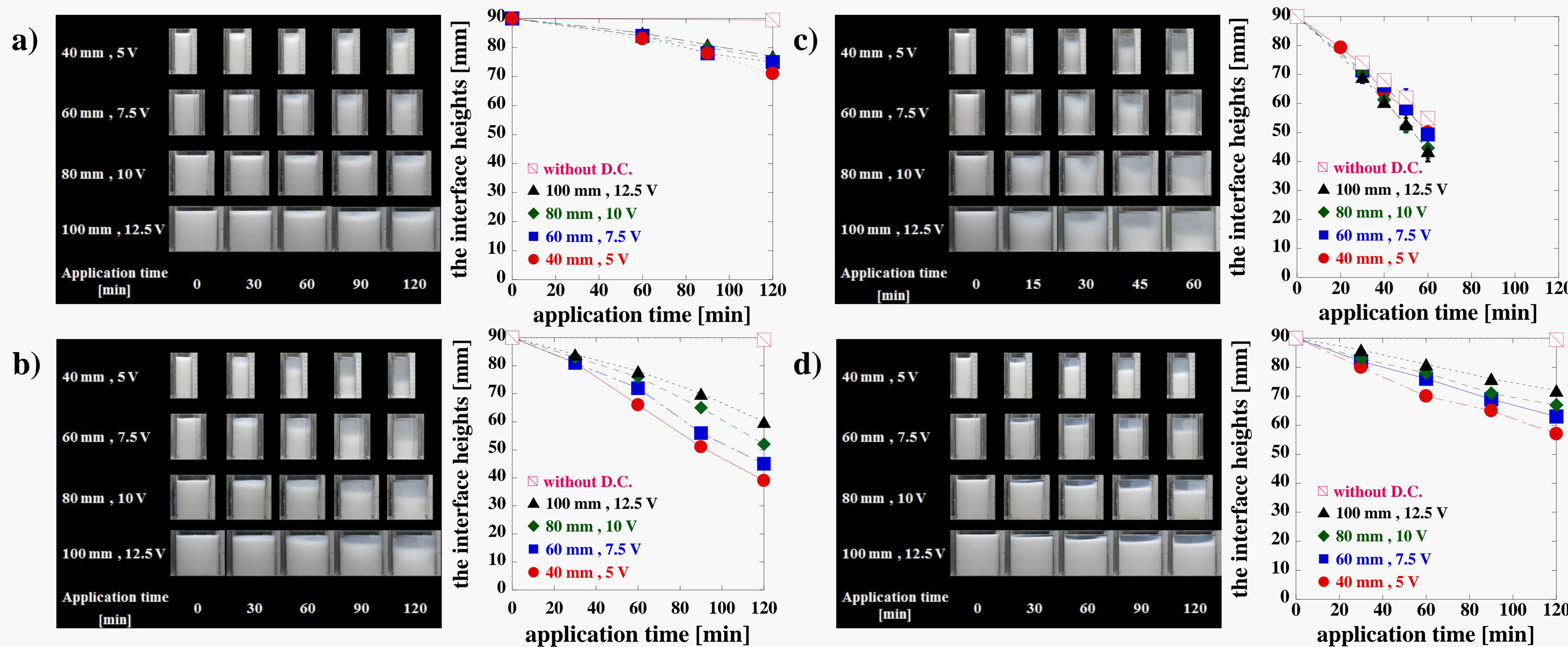


Fig.3 the change of the interface heights a) pH 3.5, b) pH 7.0, c) pH 8.8, d) pH 10.5

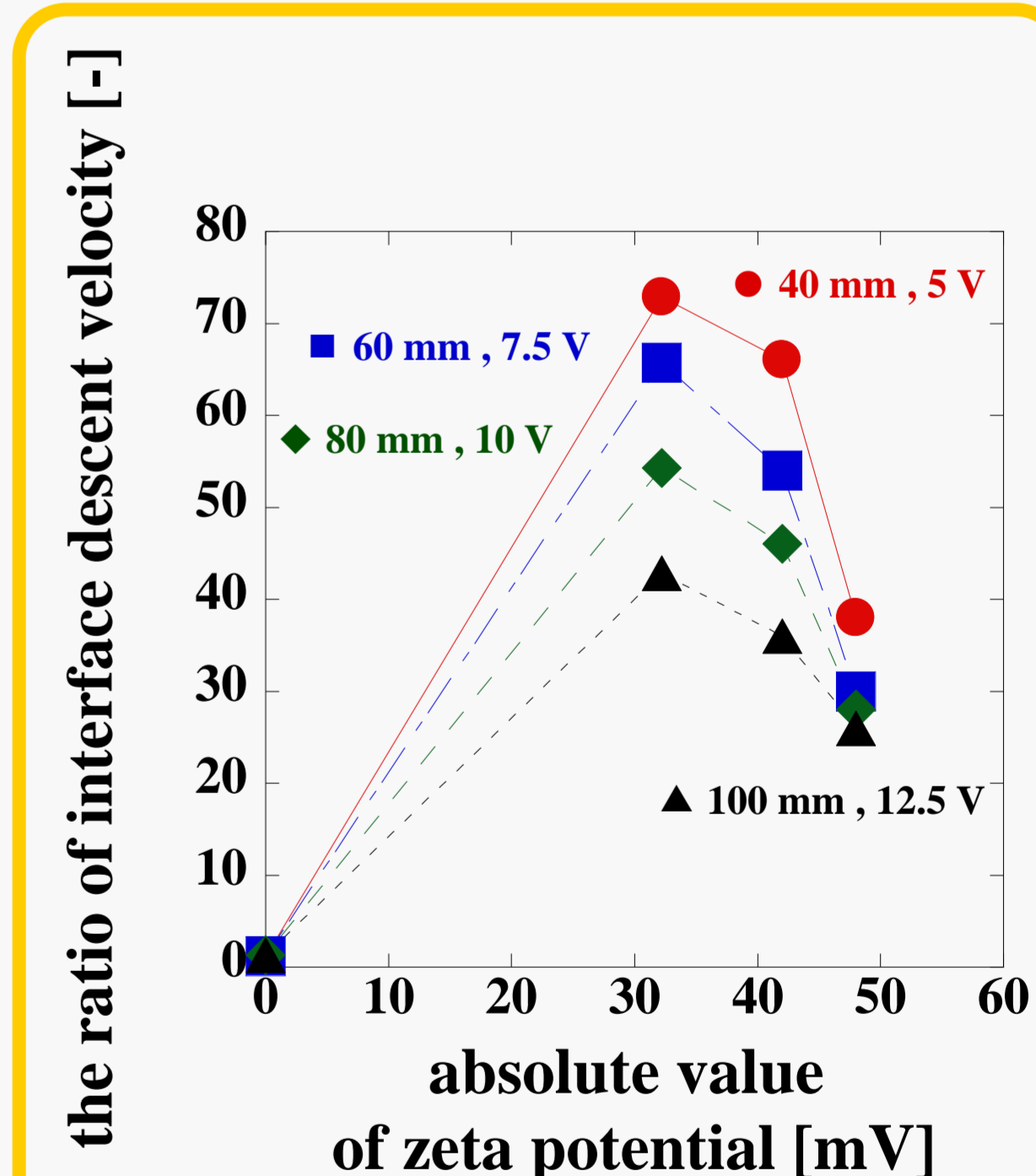


Fig.4 the aggregation efficiency

- a)  $|\zeta| = 48.3 \text{ mV}$
- b)  $|\zeta| = 32.2 \text{ mV}$
- c)  $|\zeta| = 0 \text{ mV}$
- d)  $|\zeta| = 42.6 \text{ mV}$

## Conclusion

■ In the particle aggregation using D.C. electric field, decreasing the distance between two electrodes is more effective than applying high voltage at the same electric field strength.

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